

CHEM 347 – Organic Chemistry II (for Majors)

Instructor: Paul J. Bracher

Quiz #6

(OPTIONAL)

Due in Monsanto Hall 103 by
Thursday, May 8th, 2014, 12:30 p.m.

Student Name (Printed)	
Student Signature	

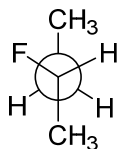
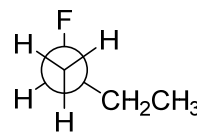
Instructions & Scoring

	Points Earned	Points Available
TOTAL		100

- Please write your answers on the official answer sheet.
- This quiz is optional. To count for a grade, your answer sheet must be turned in by the date and time listed above.
- You are allowed access to any materials you wish and may discuss the questions with other students.
- Your quiz may be photocopied.

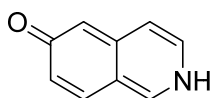
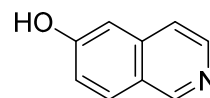
Problem I. Multiple choice (100 points total; +5 points for a correct answer, +1 point for an answer intentionally left blank, and 0 points for an incorrect answer.) For each question, select the best answer of the choices given. Write the answer, legibly, in the space provided on the answer sheet.

- (1) _____ What term best describes the relationship between compounds **A** and **B**, represented by the Newman projections below?

**A****B**

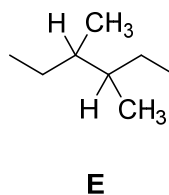
- (a) identical compounds
- (b) enantiomers
- (c) diastereomers
- (d) resonance forms
- (e) isotopes

- (2) _____ What term accurately describes the relationship between compounds **C** and **D**, whose Lewis structures are drawn below?

**C****D**

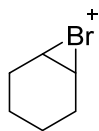
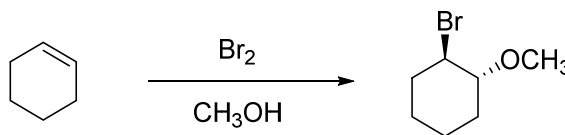
- (a) enantiomers
- (b) diastereomers
- (c) resonance forms
- (d) tautomers
- (e) conformations

(3) _____ How many stereoisomers exist of the general structure E?

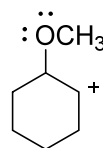


- (a) one
- (b) two
- (c) three
- (d) four
- (e) five

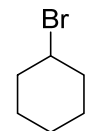
(4) _____ What structure represents an intermediate in the reaction below?



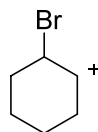
(a)



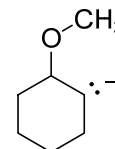
(b)



(c)

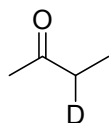
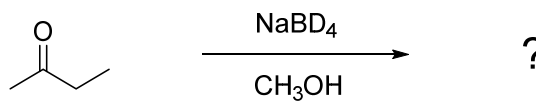


(d)

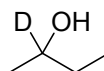


(e)

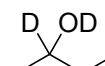
(5) _____ What is the major product expected of the following reaction?



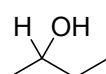
(a)



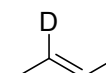
(b)



(c)

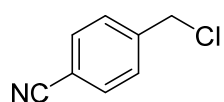
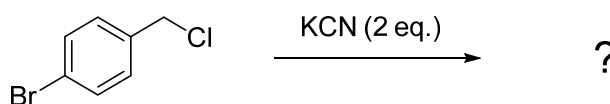


(d)

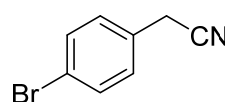


(e)

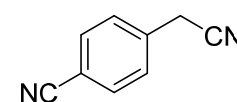
(6) _____ What is the major product expected of the following reaction?



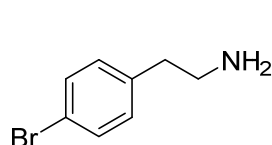
(a)



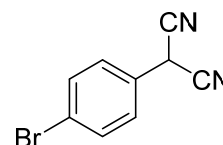
(b)



(c)

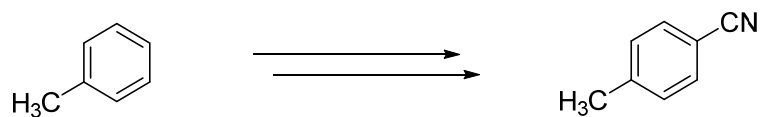


(d)



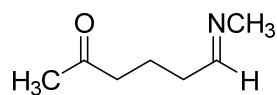
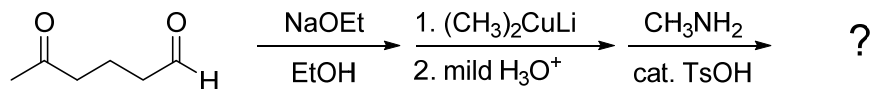
(e)

- (7) _____ What sequence of reactions is the best choice to carry out the following transformation?

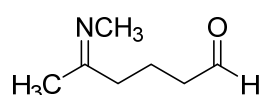


- (a) $\xrightarrow[\text{FeBr}_3]{\text{Br}_2} \xrightarrow{\text{NaCN}}$
- (b) $\xrightarrow[\text{H}_2\text{SO}_4]{\text{HNO}_3} \xrightarrow[\text{HCl}]{\text{Sn}} \xrightarrow[\text{HCl}]{\text{NaNO}_2} \xrightarrow{\text{CuCN}}$
- (c) $\xrightarrow[\text{heat}]{\text{KMnO}_4} \xrightarrow[\text{AlCl}_3]{\text{CH}_3\text{Cl}} \xrightarrow[\text{DCC}]{\text{NH}_3}$
- (d) $\xrightarrow[\text{H}_2\text{SO}_4]{\text{HNO}_3} \xrightarrow[\text{Pd-C}]{\text{H}_2} \xrightarrow{\text{CH}_2\text{O}} \xrightarrow{\text{HCl}}$
- (e) $\xrightarrow[\text{FeBr}_3]{\text{Br}_2} \xrightarrow[\text{ether}]{\text{Mg}} \xrightarrow{\text{NaCN}} \xrightarrow{\text{H}_3\text{O}^+}$

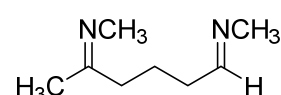
(8) _____ What is the major product expected of the series of reactions below?



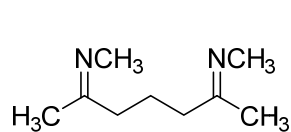
(a)



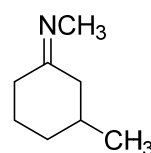
(b)



(c)

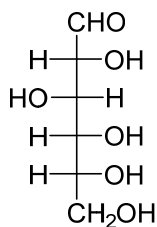


(d)

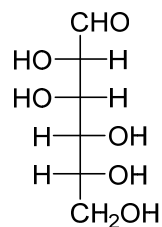


(e)

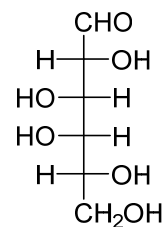
(9) _____ Which of the following carbohydrates yields an optically inactive product upon treatment with NaBH_4 ?



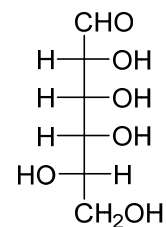
(a)



(b)



(c)

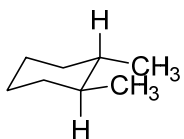


(d)

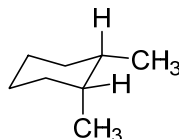
none of the above

(e)

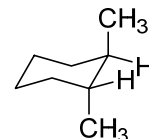
- (10) _____ Which of the following structures represents the lowest-energy conformation of (*R,R*)-1,2-dimethylcyclohexane?



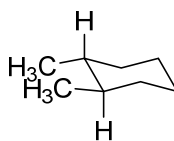
(a)



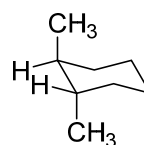
(b)



(c)

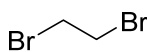


(d)



(e)

- (11) _____ Which of the following compounds will give rise to an NMR spectrum in which one of the signals is a triplet?



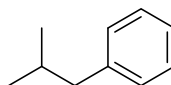
(a)



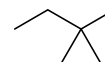
(b)



(c)

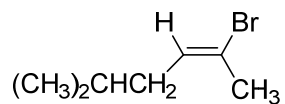


(d)



(e)

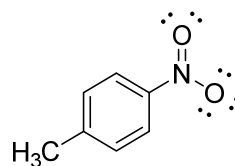
(12) _____ What is an acceptable IUPAC name for compound **F**?



F

- (a) (*Z*)-5-bromo-2-methyl-4-hexene
- (b) (*E*)-5-bromo-2-methyl-4-hexene
- (c) (*Z*)-1-bromo-1,4-dimethyl-1-pentene
- (d) (*E*)-1-bromo-1,4-dimethyl-1-pentene
- (e) (*E*)-2-bromo-5-methyl-2-hexene

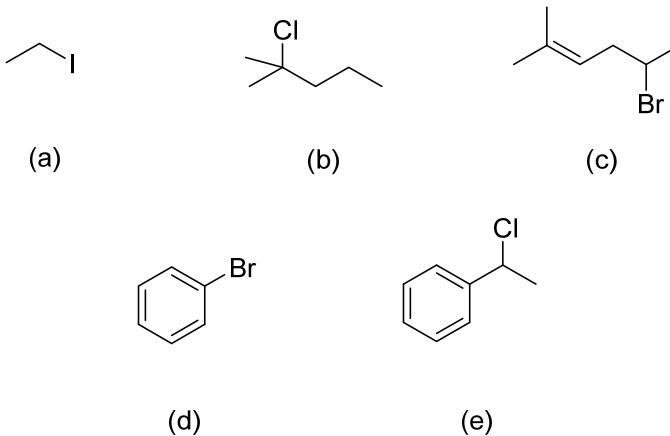
(13) _____ What is the formal charge on the nitrogen atom of *p*-nitrotoluene (**G**)?



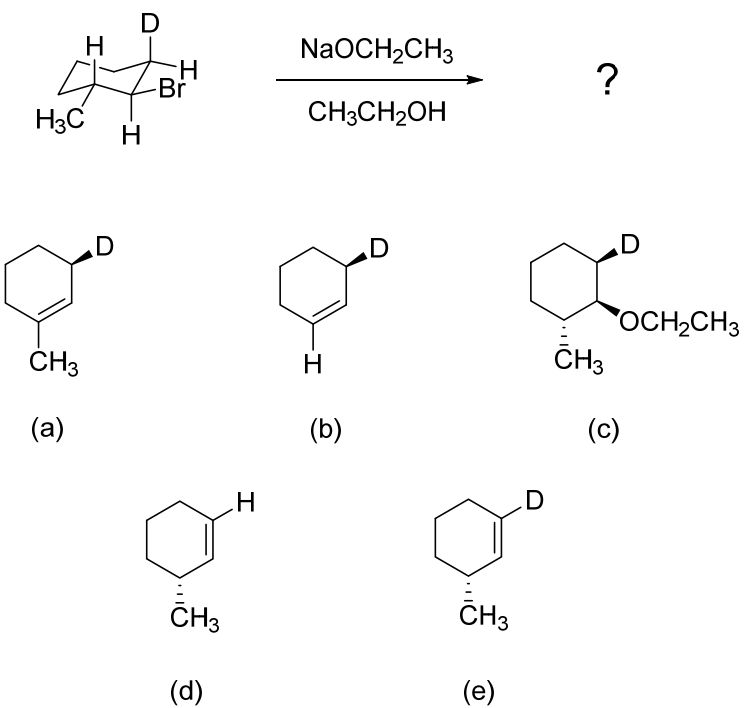
G

- (a) -2
- (b) -1
- (c) 0
- (d) +1
- (e) +2

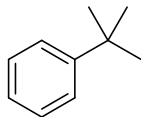
- (14) _____ Heterolytic cleavage of the carbon–halogen bond in which of the following compounds would generate a carbocation likely to undergo rearrangement?



- (15) _____ What is the major product expected of the following reaction?



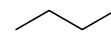
- (16) _____ Which of the compounds below reacts fastest with Br_2 in the presence of *t*-butyl peroxide?



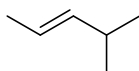
(a)



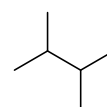
(b)



(c)

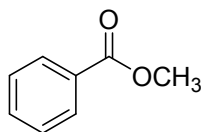
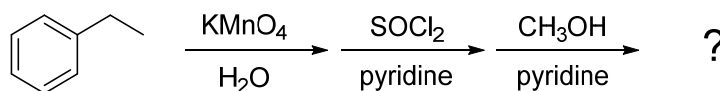


(d)

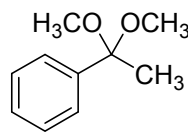


(e)

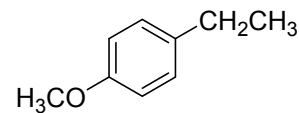
- (17) _____ What is the major product expected of the series of reactions below?



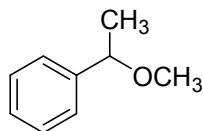
(a)



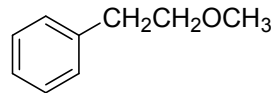
(b)



(c)



(d)

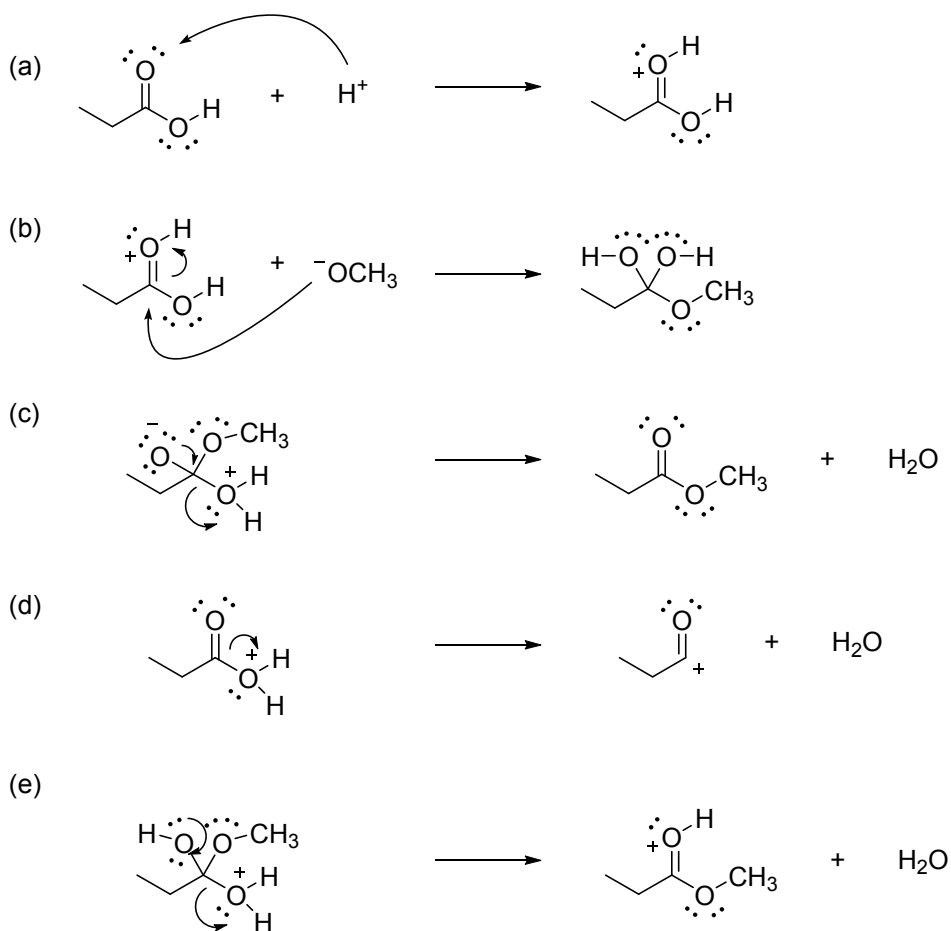


(e)

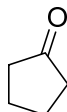
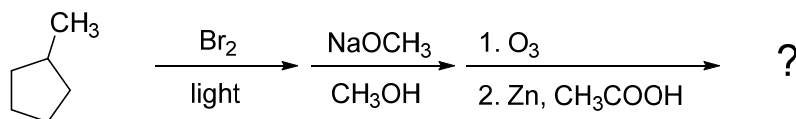
(18) _____ What rings and π bonds could be present in a stable compound with the molecular formula $C_{15}H_{25}O_3N$? (Assume each compound contains no other rings or π bonds other than those mentioned in each answer.)

- (a) a phenyl ring
- (b) a triple bond and a cyclopropane ring
- (c) two cyano (nitrile) groups
- (d) a C=C bond and two cyclohexane rings
- (e) a cyclohexene ring

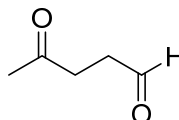
(19) _____ What choice below is a valid representation of a step in the mechanism of Fischer esterification?



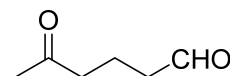
(20) _____ What is the major product expected of the sequence of reactions below?



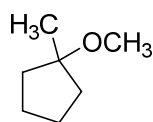
(a)



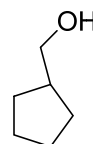
(b)



(c)



(d)



(e)