# **Section Solution Set**

### Carbonyl Chemistry III, Reactions at the Alpha Position

#### **Problem 1** Acidity of organic compounds.

a) Identify the most acidic proton in each of the following compounds.

lone pair orbital at the bridgehead would not align well with the carbonyl group

the lone pair on nitrogen donates more electron density into the carbonyl group of the amide than the oxygen lone pair donates into the carbonyl group of the ester, making the amide carbonyl less able to stabilize any negative charge density at its alpha position

b) Provide the necessary reagents to generate the enolates shown.

## **Problem 2** Propose a mechanism for this run-of-the-mill mixed Aldol condensation.

## Solution

**Problem 3** The Dieckmann reaction has a really cool sounding name. Provide a mechanism for it.

### **Solution**

FYI: